

Using and Understanding

Mathematics

A Quantitative Reasoning Approach

An aerial photograph of a hot air balloon with a purple top, blue middle, and red bottom section, floating over a golden-brown field with visible tire tracks. The balloon's shadow is cast onto the field to the left.

Third Edition

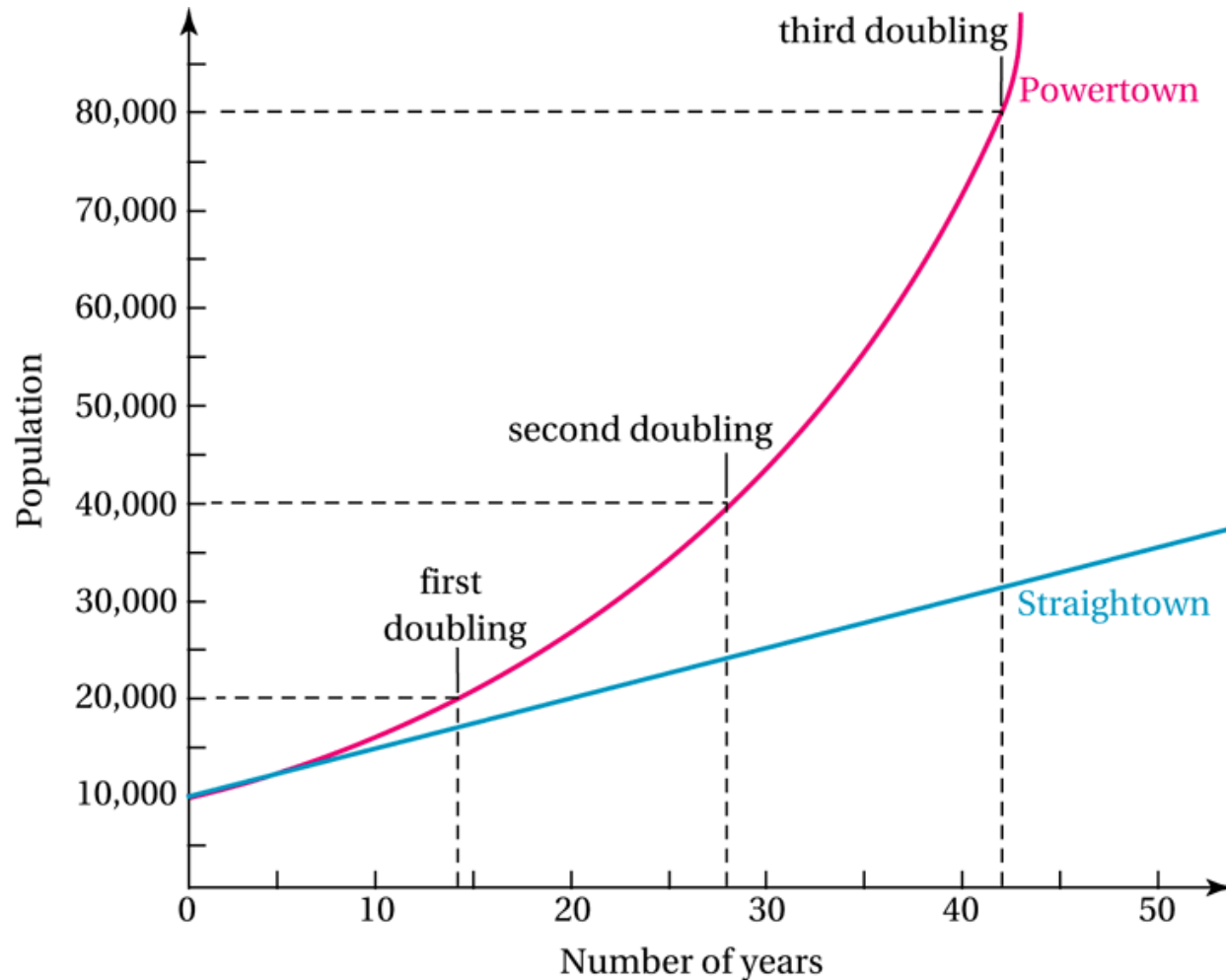
Jeffrey Bennett

William Briggs

Chapter 8



Growth: Linear versus Exponential



Growth: Linear versus Exponential

- **Linear Growth** occurs when a quantity grows by some fixed *absolute* amount in each unit of time.
- **Exponential Growth** occurs when a quantity grows by the same fixed *relative* amount—that is, by the same percentage—in each unit of time.

Key Facts about Exponential Growth

- **Exponential growth** leads to repeated doublings. With each doubling, the amount of increase is approximately equal to the sum of all preceding doublings.
- **Exponential growth** cannot continue indefinitely. After only a relatively small number of doublings, exponentially growing quantities reach impossible proportions.

Rule of 70 Approximations

- **Doubling Time**

For a quantity growing exponentially at a rate of P% per time period, the doubling time is approximately

$$T_{\text{double}} \approx \frac{70}{P}$$

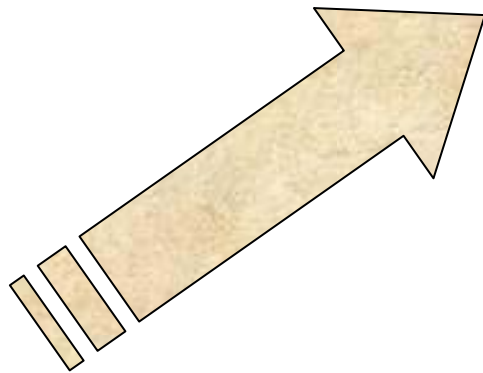
- **Half-Life Period**

For a quantity growing exponentially at a rate of P% per time period, the doubling time is approximately

$$T_{\text{half -life}} \approx \frac{70}{P}$$

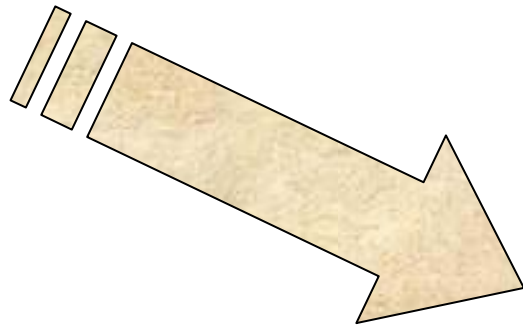
For both doubling and halving, this approximation works best for small growth rates and breaks down for growth rates over about 15%

Exponential Growth vs. Decay



Exponential Growth

$$\text{new value} = \text{initial value} \times 2^{t/T_{\text{double}}}$$



$$\text{new value} = \text{initial value} \times \left(\frac{1}{2}\right)^{t/T_{\text{half-life}}}$$

Exponential Decay

Exact Doubling Time and Half-Life Formulas

$$T_{\text{double}} = \frac{\log 2}{\log (1 + r)}$$

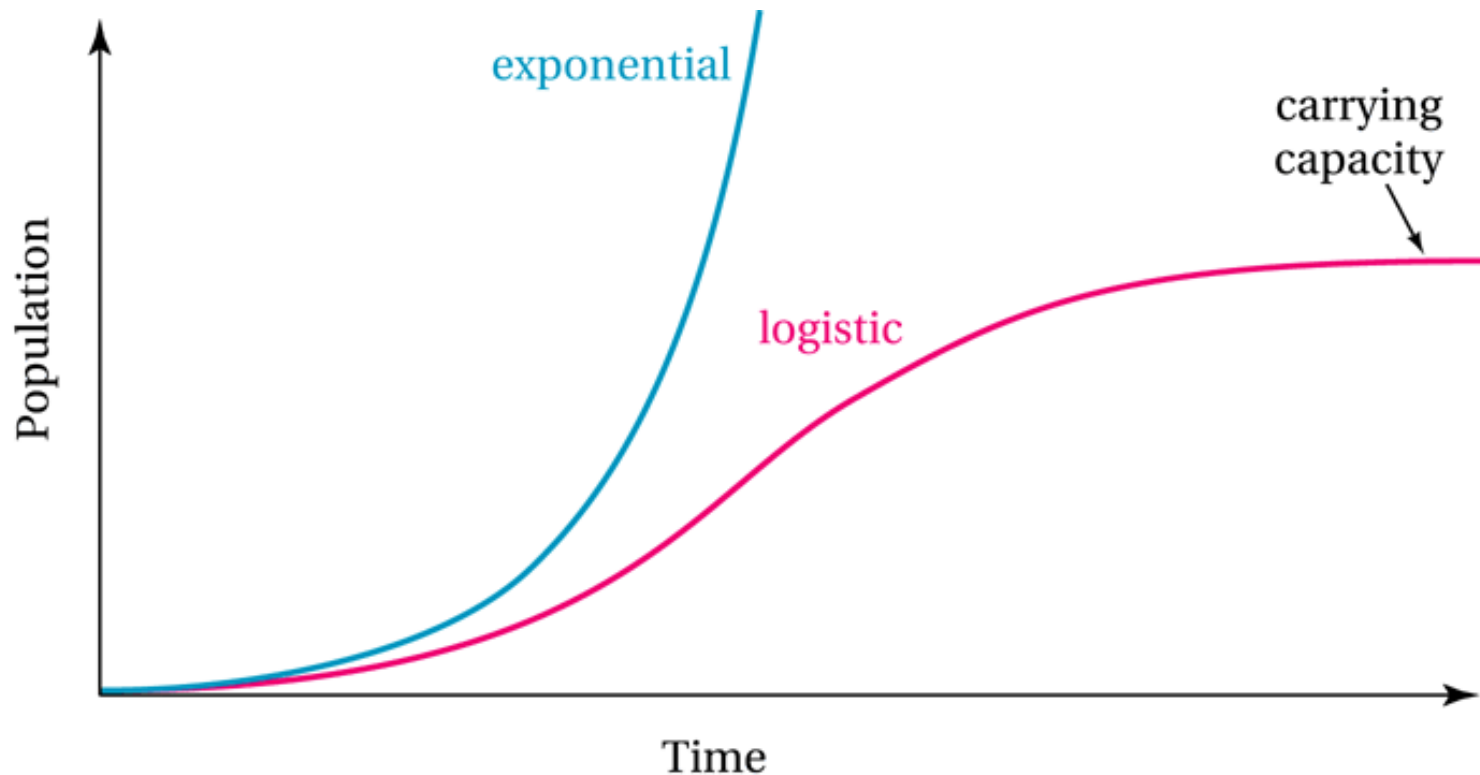
Exponential Growth
(given that $r > 0$)

$$T_{\text{half}} = \frac{-\log 2}{\log (1 + r)} \text{ or } \frac{\log (1/2)}{\log (1 + r)}$$

Exponential Decay
(given that $r < 0$)

Exponential versus Logistic Growth

8-C



Overall Growth Rate and Logistic Growth Rate

Overall Growth

The world population growth rate is the difference between the birth rate and the death rate:

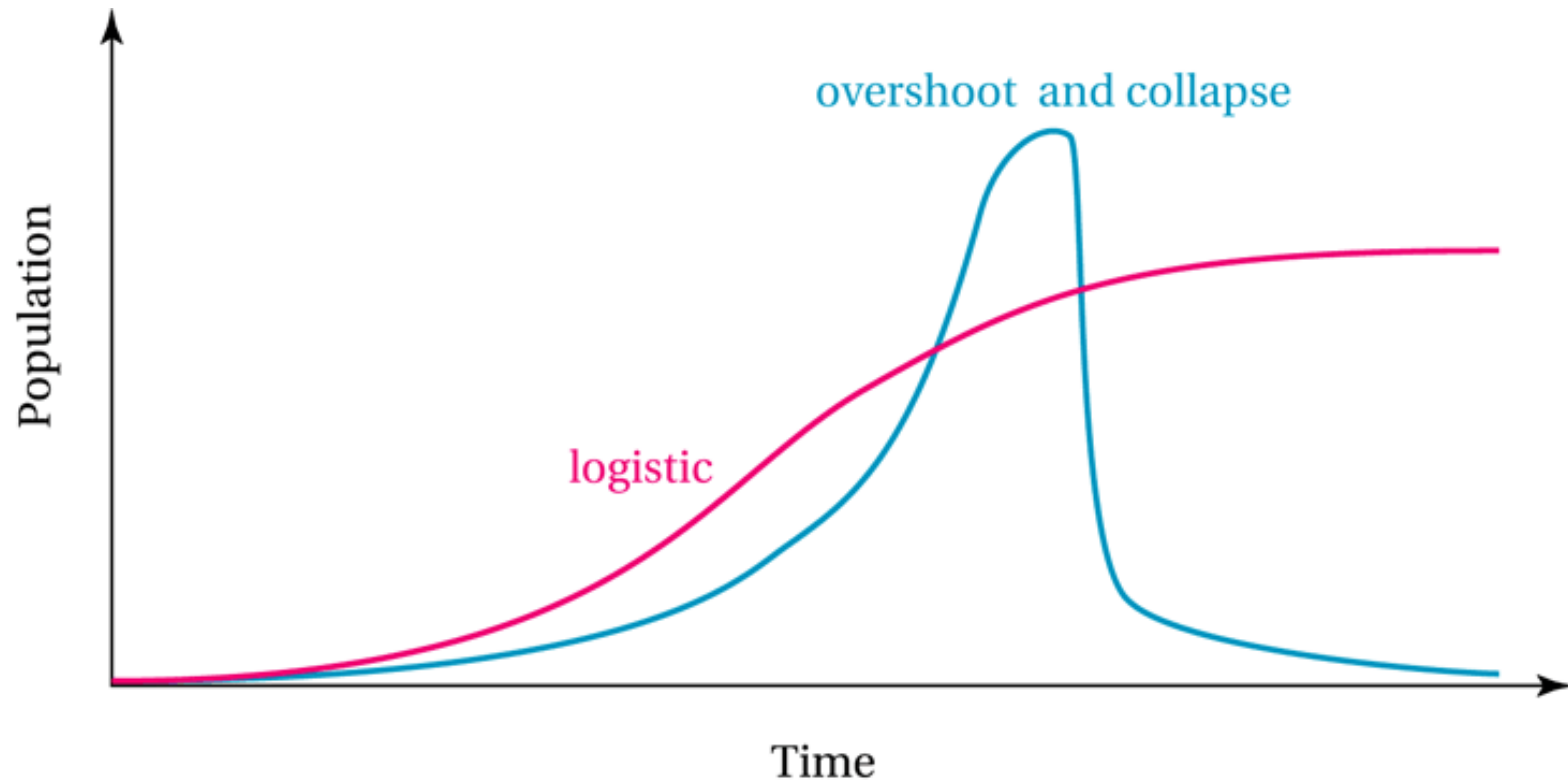
$$\text{growth rate} = \text{birth rate} - \text{death rate}$$

Logistic Growth

The logistic growth rate is dependent upon the ratio of current population to carrying capacity:

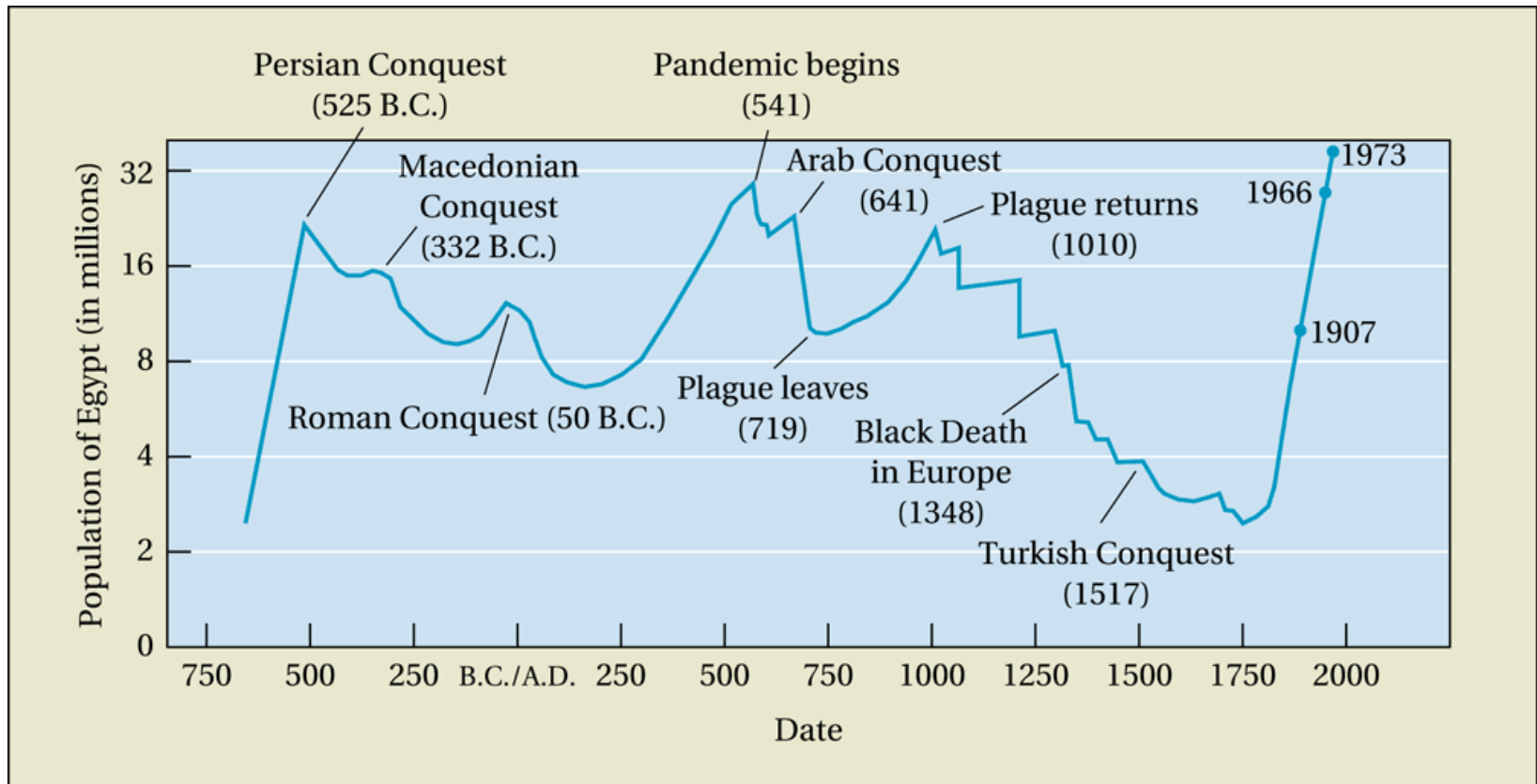
$$\text{growth rate} = \text{base rate} \times (1 - \text{population} / \text{carrying capacity})$$

Overshoot and Collapse



Difficulties of Population Prediction

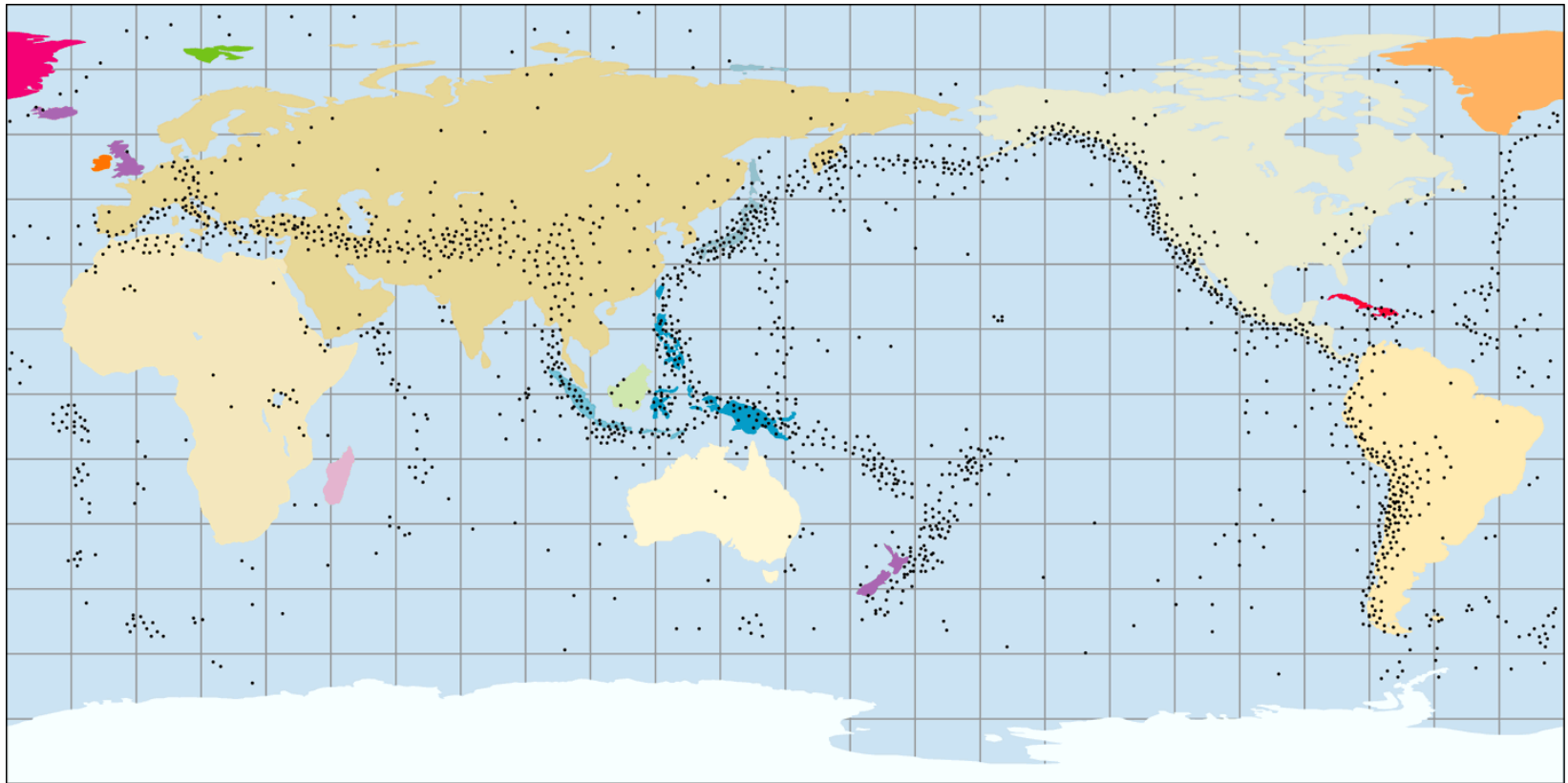
8-C



The historical population of Egypt.

SOURCE: T.H. Hollingsworth, *Historical Demography* (Ithaca, NY: Cornell University Press, 1969).

Distribution of Earthquakes



The distribution of earthquakes around the world.
Each dot represents an earthquake.
SOURCE: U.S. Geological Survey

Measuring Sound

- The **decibel scale** is used to compare the loudness of sounds.
- The loudness of a sound in decibels is defined by the following equivalent formulas:

$$\text{loudness in dB} = 10 \log_{10} \left(\frac{\text{intensity of the sound}}{\text{intensity of the softest audible count}} \right)$$

or

$$\frac{\text{intensity of the sound}}{\text{intensity of the softest audible sound}} = 10^{\frac{\text{loudness in dB}}{10}}$$

Typical Sounds in Decibels

TABLE 8.5 Typical Sounds in Decibels

DECIBELS	TIMES LOUDER THAN SOFTEST AUDIBLE SOUND	EXAMPLE
140	10^{14}	jet at 30 meters
120	10^{12}	strong risk of damage to human ear
100	10^{10}	siren at 30 meters
90	10^9	threshold of pain for human ear
80	10^8	busy street traffic
60	10^6	ordinary conversation
40	10^4	background noise in average home
20	10^2	whisper
10	10^1	rustle of leaves
0	1	threshold of human hearing
-10	0.1	inaudible sound

pH Scale

- The pH scale is defined by the following equivalent formula:

$$\text{pH} = -\log_{10}[\text{H}^+] \quad \text{or} \quad [\text{H}^+] = 10^{-\text{pH}}$$

- where $[\text{H}^+]$ is the hydrogen ion concentration in moles per liter. Pure water is neutral and has a pH of 7. Acids have a pH lower than 7 and bases have a pH higher than 7.